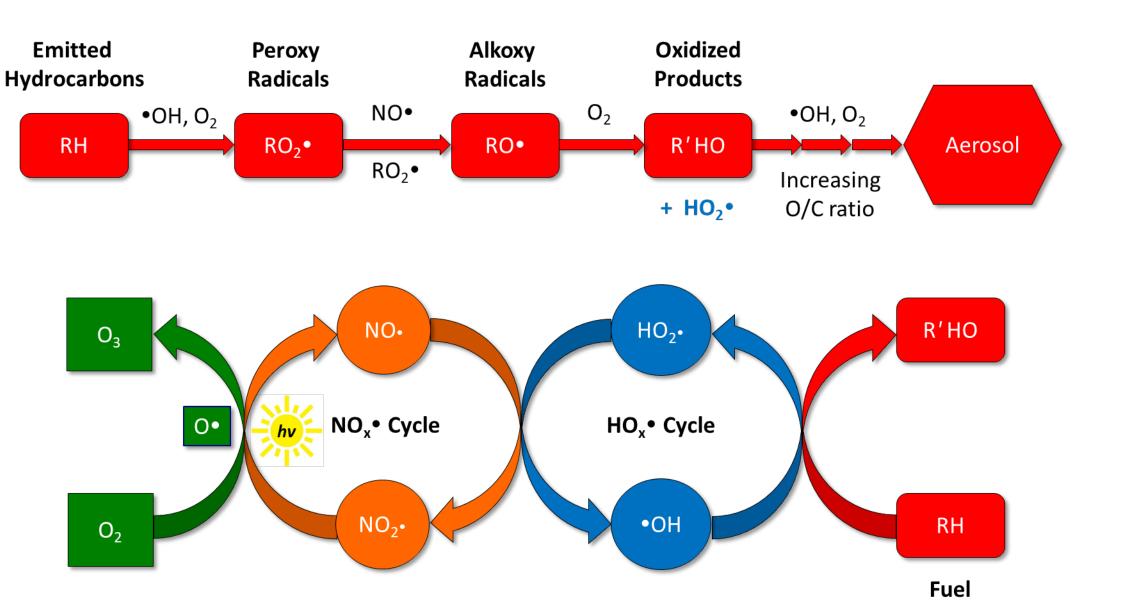
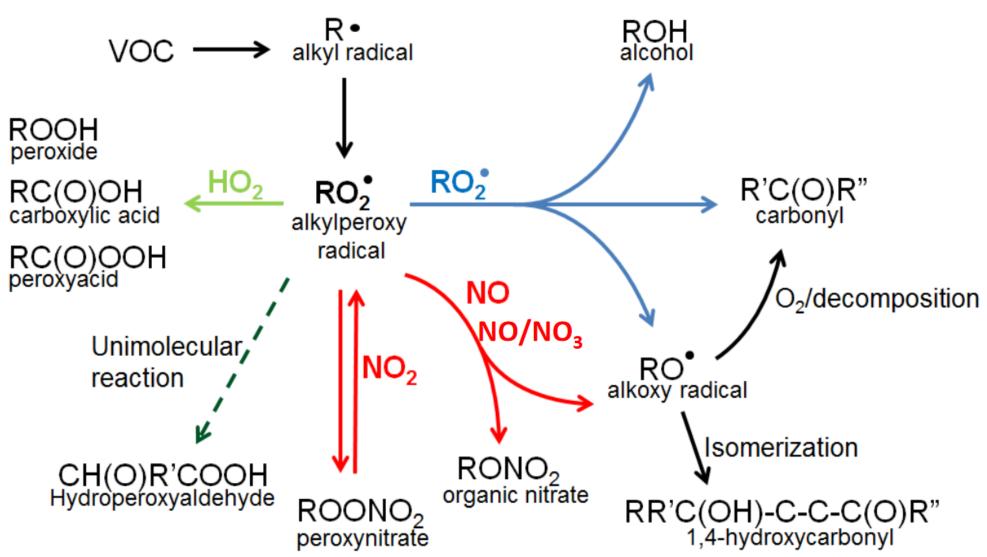
Kinetics and Spectroscopy via Multiplexed Photoionization Mass Spectrometry Okumura Group

Chemistry of Air Pollution

- Organic peroxy radicals play a central role in the chemistry of the atmosphere.
- Chain reactions of these radicals form a variety of highly oxygenated moieties, which condense to form aerosols, with significant impact on human health and climate
- Reactions with NO and HO₂ drive photochemical production of ozone in troposphere, a main component of urban smog

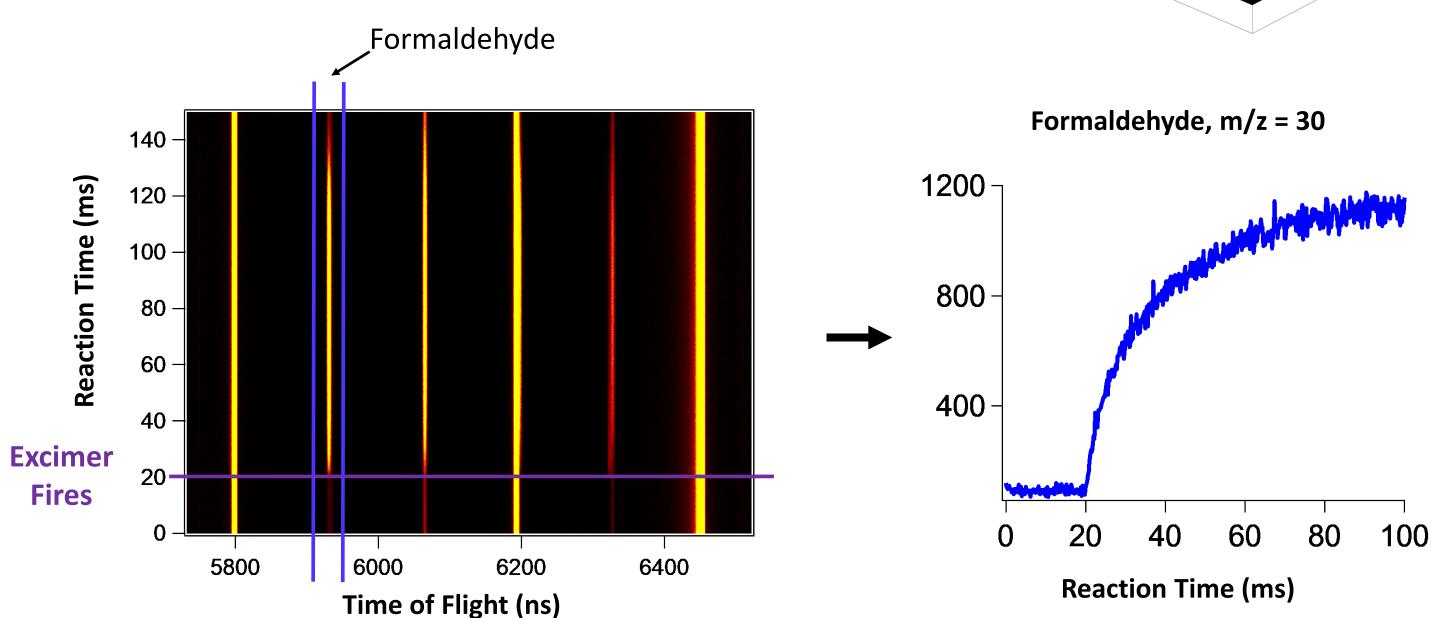


Complex interrelationships make isolating individual reactions difficult



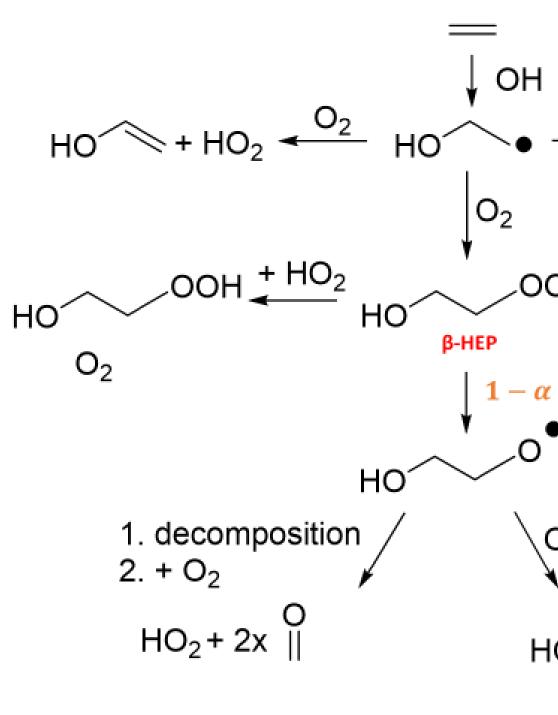
Our approach: multiplexed photoionization mass spectrometry

- Collect full mass spectrum every 20 µs over the course of 0 - 150 ms, giving kinetics as a function of mass
- Tunable photon energy from synchrotron at Lawrence Berkeley Labs yields selective ionization of same-mass species

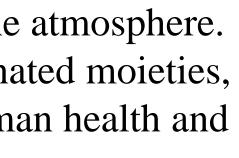


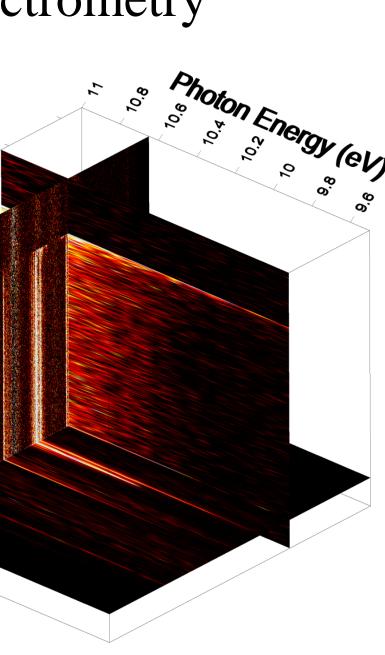
Peroxy Radical Kinetics

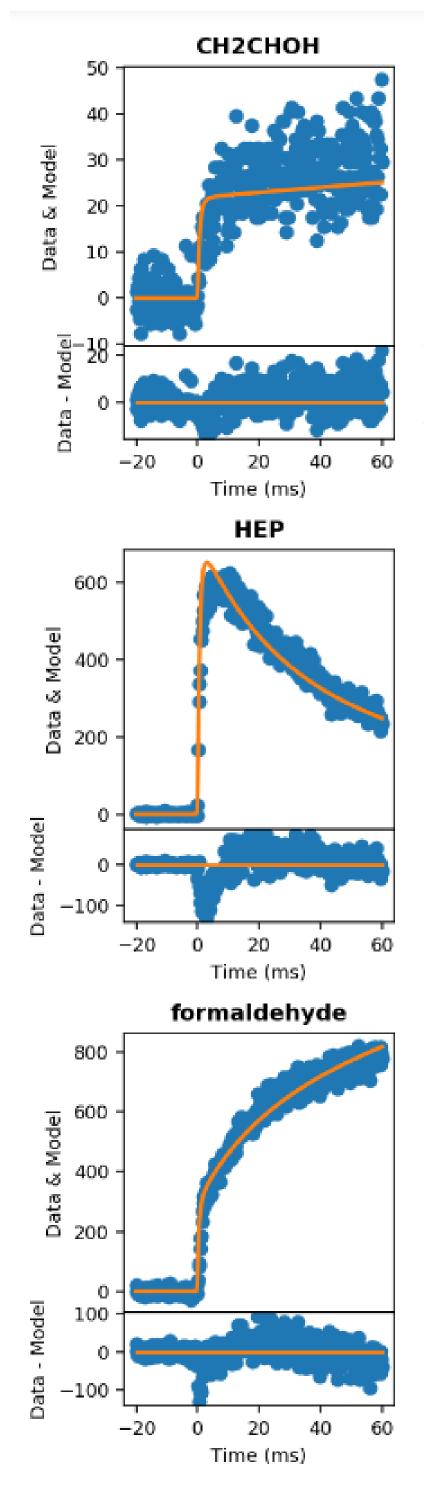
- β -HEP serves as a model system for OH-initiated oxidation of alkenes
- It was reported that its self reaction was enhanced in the presence of water through a H-bonded complex
- Only disappearance of β -HEP observed; effects on branching fraction α unknown

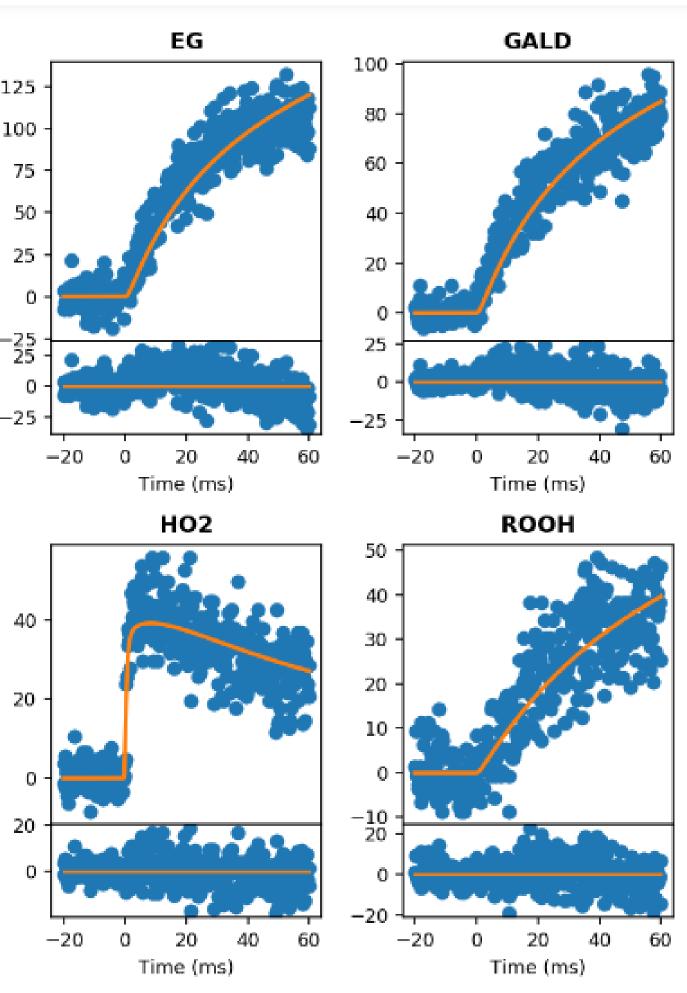


- Using MPIMS, we can observe all species except $HOCH_2CH_2$ and HOCH₂CH₂O·
- By varying water concentrations and fitting to a kinetic model, branching fractions can be determined







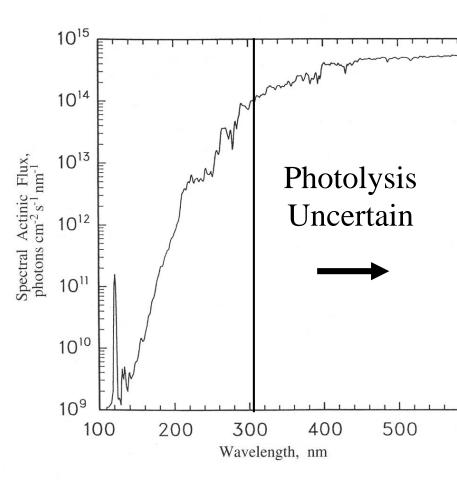


$$O_2 \rightarrow OH + 2x \parallel^{O}$$

 $HO \rightarrow OH + 0$
 $HO \rightarrow OH + 0$

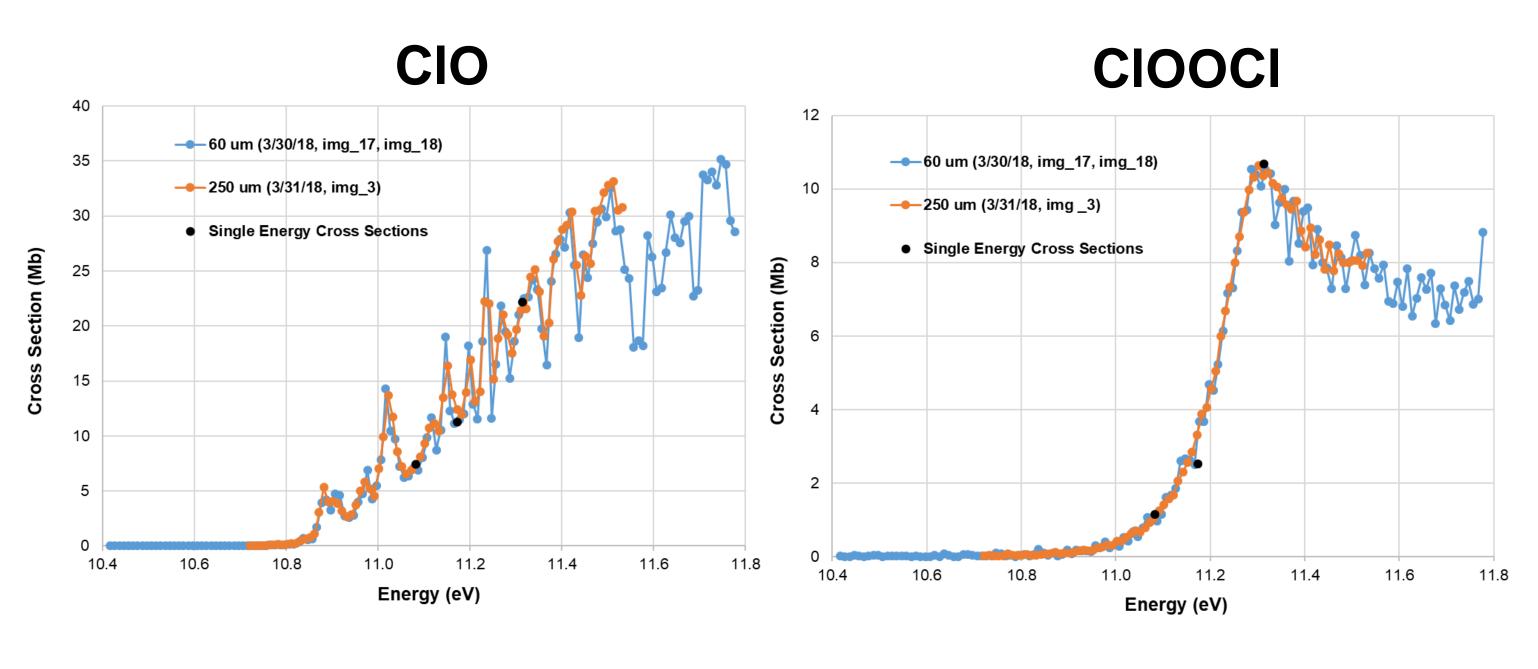
 $\sim 0 + HO_2$

Initial fit of model to data



Strategy: produce ClOOCl from ClO + ClO and measure depletion/products of ClOOCl photolysis longwave of 308 nm

to put concentrations on absolute scale



- Photolysis cross section of ClOOCl
- high-harmonic generation of an ultrafast laser
- NSF Graduate Research Fellowship Program
- NASA Earth and Space Science Fellowship
- NSF Award # 1413712

Ozone Depletion Chemistry

• ClO_x chemistry important in the stratosphere, especially in ozone destruction • Photolysis products never measured for $\lambda > 308$ nm

> $ClO + ClO + M \rightarrow ClOOCl + M$ $ClOOCl + hv \rightarrow ClOO + Cl$ $ClOO \rightarrow Cl + O_2$ $2 \times (Cl + O_3 \rightarrow ClO + O_2)$ Net: $2 O_3 \rightarrow 3 O_2$

Currently measuring photoionization cross section of ClO, ClOOCl

Photoionization spectrum of ClO shows structure from Rydberg autoionization resonances \rightarrow fundamental chemical physics

Future Work

• Photoproduct branching ratios for ClOOCl photolysis • Construction of photoionization mass spectrometer at Caltech using

Funding